SI Physics - Full Discipline Demo

Introduction to Experimental Errors and Uncertainty

Final Report - Answer Guide

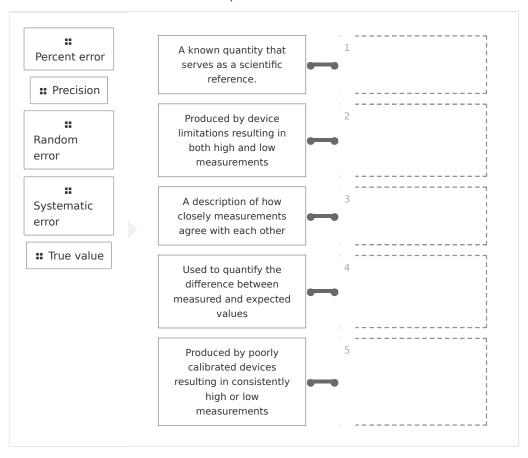
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Test Your Knowledge



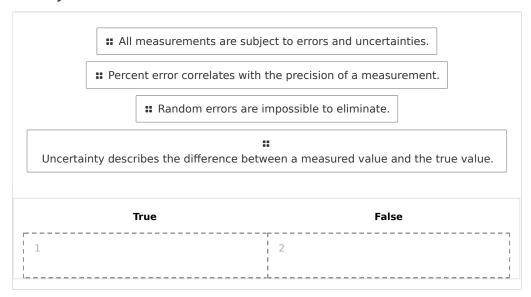
Match each term with the best description.



Correct answers:

- 1 True value 2 Random error 3 Precision 4 Percent error
- 5 Systematic error

Identify each statement as true or false.



Correct answers:

All measurements are subject to errors and uncertainties.

Random errors are impossible to eliminate.

2 Percent error correlates with the precision of a measurement.

Uncertainty describes the difference between a measured value and the true value.

Exploration

-	describes the range of values within which the true value lies.	
	Accuracy	
	Error	
	Uncertainty	~
_	All of the above	



	an object.
	O True
	○ False
	Percent uncertainty is a measure of the of a device.
	○ accuracy
	○ precision ✓
	systematic error
	 All of the above
	Accuracy is a description of how close a measured value is to the true value, and is quantified by percent error. Accuracy is related to the systematic error in measurements, and is influenced by device calibration and reading techniques. Precision describes how close two or more independent measurements agree with each other, and is quantified by percent uncertainty. Precision is related to the random error in measurements, and is influenced by device limitations.
	tematic and/or random error present in your measurements? Explain your answer by ing Data Table 1.
systema	ts should find evidence of both error types in Data Table 1. HOL testing suggests that atic error was present as all measurements were greater than the true value. Random error esent in HOL data as measurements differed from each other.



Describe potential sources of systematic error and random error in your measurements. Explain your answer.	
Student answers will vary but should include systematic sources of improperly calibrated stopwatch and measuring tape and the recording technique that caused lag between starting the watch and releasing the ball, and between hearing the contact and stopping the watch. Random error sources include limitations of the stopwatch, operator error, and differences between how each person performed their assigned tasks in the trials.	
Did repeating trials improve the accuracy and precision of your measurements? Explain yo	ur
Student answers will vary but should reference the percent error and percent uncertainty values	

Data Table 1: Time for Object to Fall 1.00 m (SAMPLE ANSWER BELOW) $\,$

	Trial 1	Trial 2	Trial 3	Trial 4
Student Operator	Student answers will vary for data table. 0.48	0.47	0.46	0.49
Assistant Operator	0.51	0.52	0.49	0.53
Trial Average and Uncertainty	0.50 +/- 0.02 s	0.50 +/- 0.03 s	0.48 +/- 0.02 s	0.51 +/- 0.02 s
Percent Uncertainty	4.0%	6.0%	4.2%	3.9%
Percent Error	10.6%	10.6%	6.2%	12.8%

calculated from each trial. Accuracy is unlikely to improve over repeated trials since it is related to

systematic error. Students may, however, find increased precision in repeated trials.

Competency Review

is the difference between a measured value and the true value.	
○ Error	~
Precision	
Uncertainty	
 All of the above 	
Improperly calibrated devices cause error.	
precision	
random	
systematic	✓
All of the above	
Random error results in measurements with unpredictable high and low values.	
○ True	✓
- False	
Accuracy is quantified as	
percent error	✓
percent uncertainty	
random error	
All of the above	
Low percent error implies that measurements are closely grouped.	
○ True	
False	~
I and the second	



O 1.47%		
○ 2.02%		
2.94%		
3.35%		
The length of an object is measured two time 1.86 m. The percent uncertainty of the meas		
3.74%	✓	
3.88%		
0 7.20%		
0 17.50%		
The quality of measurement data with high percent error can be improved with	percent uncertainty but low	
	percent uncertainty but low	
percent error can be improved with	percent uncertainty but low	
ercent error can be improved with a correction factor	percent uncertainty but low	

Extension Questions

Jason measures the mass of a sample and finds it to be 12.21 g. He states in his lab report that the measurement is both accurate and precise. Explain why his statement is incorrect and how it could be corrected. (SAMPLE ANSWER BELOW)

Jason cannot state that his measurement is accurate and precise without quantifying accuracy and precision by providing the percent error and percent uncertainty of his measurement. Jason should compare his sample to the known value to quantify accuracy and systematic error in his measurement. Furthermore, he should compare his measurement to other measurements of the sample to quantify precision and random error in his measurement. Lastly, Jason should state the level of uncertainty in his measurement as 12.21 +/- ____ g.

