SI Geology - Full Discipline Demo

Crystallization

Final Report - Answer Guide

InstitutionScience Interactive UniversitySessionSI Geology - Full Discipline DemoCourseSI Geology - Full Discipline Demo

Instructor Sales SI Demo

Test Your Knowledge



Classify the following statements as true or false.

True False

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Crystal formation through solid state diffusion occurs when atoms that are dissolved in water bond together and separate from the solution.

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Crystal formation due to precipitation from a gas primarily occurs around volcanic vents and geysers.

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Biomineralization occurs when living organism cause minerals to precipitate within or adjacent to an organism's body.

- : Low pressure conditions during crystal formation will result in large crystals.
- # The size and quantity of crystals formed increases in slightly acidic environments.

Correct answers:

1

Crystal formation due to precipitation from a gas primarily occurs around volcanic vents and geysers.

Biomineralization occurs when living organism cause minerals to precipitate within or adjacent to an organism's body.

The size and quantity of crystals formed increases in slightly acidic environments.

2

Crystal formation through solid state diffusion occurs when atoms that are dissolved in water bond together and separate from the solution.

Low pressure conditions during crystal formation will result in large crystals.

Exploration



Which of the following is not a method of crystal formation?	
○ Liquid-State Diffusion ✓	
Precipitation from a Solution	
 Biomineralization 	
 Precipitation of a Gas 	
When a solution is cooled rapidly, crystals grow a solution cooled slowly.	
slower than	
○ faster than	
at the same rate as	
Exercise $oldsymbol{1}$	
2. Identify the environmental factors that could affect crystal growth and describe hor these factors influence crystal growth.	ow each

3. How did your theories about crystalization patterns under the different termperatures compare to your results? If you were unable to grow any of the crystals in any of the conditions, explain why this could have happened.

Data Table 1: Saturation of the Solutions (SAMPLE ANSWER BELOW)

Chemical Compound	Weight of Cup (g)	Starting Weight (g)	Final Weight (g)	Amount Used for Satur
$MgSO_4$	3.2	50	7.1	42.9
$Al_2(SO_4)_3$	3.3	25	5.6	19.4
CuSO ₄	3.3	25	5.8	19.2

Data Table 2: Theorized Crystal Growth

(SAMPLE ANSWER BELOW)

Chemical Compound	Observations: Room Temperature	Observations: Refrigerated Crystals
$MgSO_4$	Student answers will vary	Student answers will vary
$Al_2(SO_4)_3$	Student answers will vary	Student answers will vary
CuSO ₄	Student answers will vary	Student answers will vary

Data Table 3: Crystal Growth Observations

(SAMPLE ANSWER BELOW)

Chemical Compound	Observations: Room Temperature	Observations: Refrigerated Crystals
$MgSO_4$	Observations will vary. Size should be larger than the refrigerated crystals.	Size should be smaller than the room temperature crystals.
$Al_2(SO_4)_3$	Size should be larger than the refrigerated crystals.	Size should be smaller than the room temperature crystals.
CuSO ₄	Size should be larger than the refrigerated crystals.	Size should be smaller than the room temperature crystals.

Photo 1: MgSO4 Room Temperature (SAMPLE ANSWER BELOW)





Photo 2: MgSO4 Refrigerated (SAMPLE ANSWER BELOW)







Photo 3: AI2(SO4)3 Room Temperature (SAMPLE ANSWER BELOW)

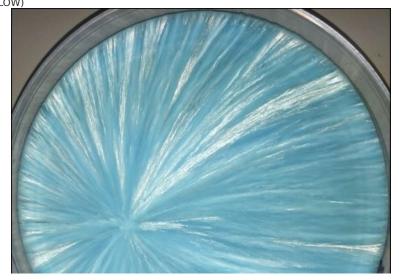






Photo 4: Al2(SO4)3 Refrigerated (SAMPLE ANSWER BELOW)

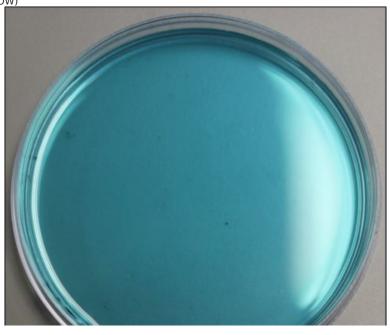
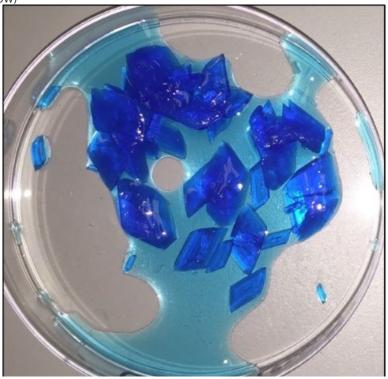






Photo 5: CuSO4 Room Temperature (SAMPLE ANSWER BELOW)









Competency Review

which of the following is a method of crystal destruction?	
Solid-State Diffusion	
Precipitation from a Solution	
Dissolution	✓
Biomineralization	
Minerals form through precipitation of a gas occurs when hot gases or steam enter the atmosphere and abruptly cool.	
○ True	✓
• False	
When fluid availability is increased from higher water content, crystal growth is promoted under longer geologic time scales.	
○ True	
○ False	✓



	hich of the following environmental conditions will result in smaller ystal formation?	
	Low Temperature	
	Low Pressure	✓
	Acidic Environments	
	 Slow Cooling 	
of pl te cr	student is growing crystals for her geology lab. She saturates a solution aluminum sulfate and divides the solution in half into two beakers. She aces one beaker in the refrigerator and one on a desk at room mperature. Which beaker can Michele expect to produce the smaller ystals? The beaker on the desk at room temperature The beaker in the refrigerator	
in	forms crystals when atoms or ions move through a solid to arrange to a new crystal structure.	
	Precipitation	
	Biomineralization	
	BiomineralizationMelting	

Extension Questions

Describe how a single large crystal could be grown from a saturated $CuSO_4$ solution. Include an explanation of how the crystal continues to grow. (SAMPLE ANSWER BELOW)

In order to get one single crystal, a seed crystal is needed from an original batch of grown crystals. The seed crystal is then placed in a beaker of only saturated solution (no other crystals should be present), and the molecules from the solution will be attracted to the seed crystal and continue to add to the size. The beaker should be cleared of any small crystals that form so that the molecules will centralize on the seed crystal.

